

DESIGN OF MICROWAVE DIELECTRICS BASED ON CRYSTALLOGRAPHY

Hitoshi OHSATO

Nagoya Industrial Science Research Institute, Nagoya 464-0819, Japan
Nagoya Institute of Technology, Nagoya 466-8555, Japan

ABSTRACT

The authors have been studying correlation between crystal structure and microwave dielectric properties based on crystallography. New dielectric materials were designed based on the origins of the properties clarified as presented in following three categories. (I) Low loss microwave dielectrics designed by low internal strain due to compositional ordering, perfect crystallinity without defects and impurities, and high symmetry and high crystallographic densities. (II) Dielectric constants due to large unit cell with inversion symmetry i , large rattling factor accompanying expanding polyhedron. On the other hand, low dielectric constant due to tight polyhedron due to covalence such as silicates. (III) Low temperature coefficients of resonant frequencies (TCf) are affected by the tilted octahedra depending on the crystal transitions. The TCf is designed generally by combination positive and negative TCf , which might be clarified the origin of TCf . In this paper, some examples with relationship between crystal structure and microwave dielectric properties are presented, and origin of the properties are clarified.

INTRODUCTION

The authors have been studying correlation between crystal structure and microwave dielectric properties based on crystallography¹⁻³⁾. Microwave dielectrics are expected following properties: high quality factor Q for resonate with microwave, dielectric constant ϵ_r of high for shortage wavelength, and low for millimeter wave, and near zero temperature coefficient of resonate frequency TCf for stability usage on wide temperature range. These properties are depending on the crystal structure. So, after relationship between crystal structure and properties has been studied and the origin of the properties is clarified, new dielectrics with high properties have been designed.

New dielectric materials improved the properties were designed based on the origins of the properties clarified as presented in following three categories. (I) Low loss microwave dielectrics designed by low internal strain due to compositional ordering, perfect crystallinity without defects and impurities, and high symmetry and high crystallographic densities. (II) High dielectric constants due to large unit cell with inversion symmetry i , accompanying large rattling factor on the expanding polyhedron. On the other hand, low dielectric constant due to tight polyhedron due to covalence such as silicates. (III) Near zero temperature coefficient of resonant frequency TCf designed generally by combination positive and negative TCf which has

clarified the origin of TCf .

In this paper, examples with relationship between crystal structure and microwave dielectric properties of the microwave dielectrics are presented, and origins of the properties are clarified.

EXPERIMENTAL

These compounds are almost fabricated by solid state reactions, identified and obtained the lattice parameters by X-ray powder diffraction (XRPD)⁴⁾. The crystal structures were analyzed by RADY program⁵⁾ for single crystal, or by Rietveld method⁶⁾ for powder patterns. And the microwave dielectric properties were evaluated by Hakki and Coleman's method⁷⁻⁸⁾, as presented previous papers¹⁻²⁾.

RESULTS AND DISCUSSIONS

Three microwave dielectric properties are presented in the order of High Q , ϵ_r and TCf .
I) High Q

I-1) (a) High Q by compositional ordering

The pseudo-tungstenbronze $Ba_{6-3x}R_{8+2x}Ti_{18}O_{54}$ (R = rare earth) solid solutions shows the highest Qf value at $x = 2/3$ as shown in Fig. 1(a)^{1,9)}, which internal strain is the smallest at the composition as shown in Fig. 1(b).

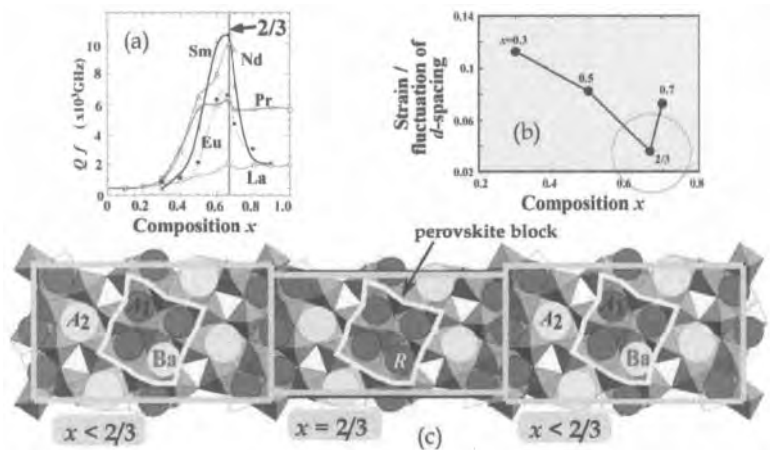


Fig. 1. (a) The Qf of pseudo-tungstenbronze $Ba_{6-3x}R_{8+2x}Ti_{18}O_{54}$ (R = Sm, Nd, Pr, Eu and La) as a function of composition x . (b) Strain/fluctuation of d -spacing of the Sm-analogous as a function of x . (c) Crystal structure with different occupation in the perovskite blocks in $x < 2/3$ case.

At the $x = 2/3$ composition of the solid solutions, the crystal structure show compositional ordering for R ions occupying A_1 site in perovskite block as shown in a part of $x = 2/3$ on Fig. 1(c). At the $x < 2/3$, Ba ions occupy statistically at A_1 sites in the perovskite block such as $[R_{8+2x}Ba_{2-3x}V_x]_{A1}$ as shown in a part of $x < 2/3$ on Fig. 1(c). The internal strains as shown in Fig. 1(b) are explained as the fluctuation of d -spacing of lattice constants. At $x < 2/3$, the fluctuation of d -spacing becomes large depending on the statistical occupation of Ba ions in perovskite blocks. On the other hand, at $x = 2/3$, the fluctuation of d -spacing becomes to be reduced depending on the all unit cells with the same size.

(b) Design more compositional ordering on the pseudo-tungstenbronze compounds

Low Qf value of ca. 200 GHz on the $Ba_6R_8Ti_{18}O_{54}$ ($R = Nd$) composition in the vicinity of $x = 0$ for $Ba_{6-3x}R_{8+2x}Ti_{18}O_{54}$ as shown in Fig. 1(a) was improved to 6,000 GHz by substitution of Sr for Ba in the perovskite blocks of $[Nd_8Ba_{2-\alpha}Sr_\alpha]_{A1}[Ba_4]_{A2}Ti_{18}O_{54}$ as shown in Fig. 2⁽⁹⁾. The substitution of Sr with smaller ionic radius than Ba ion must reduce the internal strain decreasing the fluctuation of d -spacing. This substitution of Sr for Ba also introduces a kind of the compositional ordering in A_1 and A_2 sites.

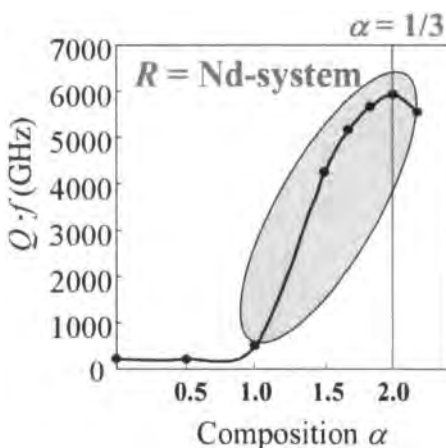


Fig. 2. Qf values of the Nd-analogy with $x = 0$ were improved from 200 to 6,000 GHz by substitution of Sr for Ba⁽⁹⁾.

I-2) High Q by high symmetry

As presented at previous section, compositional ordering on the pseudo-tungsten bronze solid solutions brings high Q . And many researchers presented ordering of complex perovskite such as $Ba(Mg_{1/3}Ta_{2/3})O_3$ (BMT), $Ba(Zn_{1/3}Ta_{2/3})O_3$ (BZT) and $Ba(Zn_{1/3}Nb_{2/3})O_3$

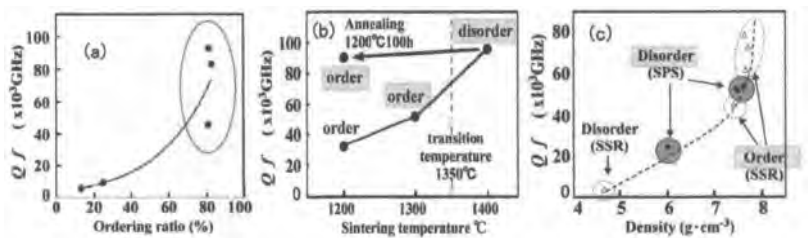


Fig. 3. (a) Qf of BZT ceramics as a function of ordering ratio, (b) Qf of BZN with order-disorder phase transition at 1350 °C as a function of sintering temperature, (c) Qf of BZT as a function of density by solid state reaction (SSR) and spark plasma sintering (SPS). Order: ordered perovskite, Disorder: disordered perovskite.

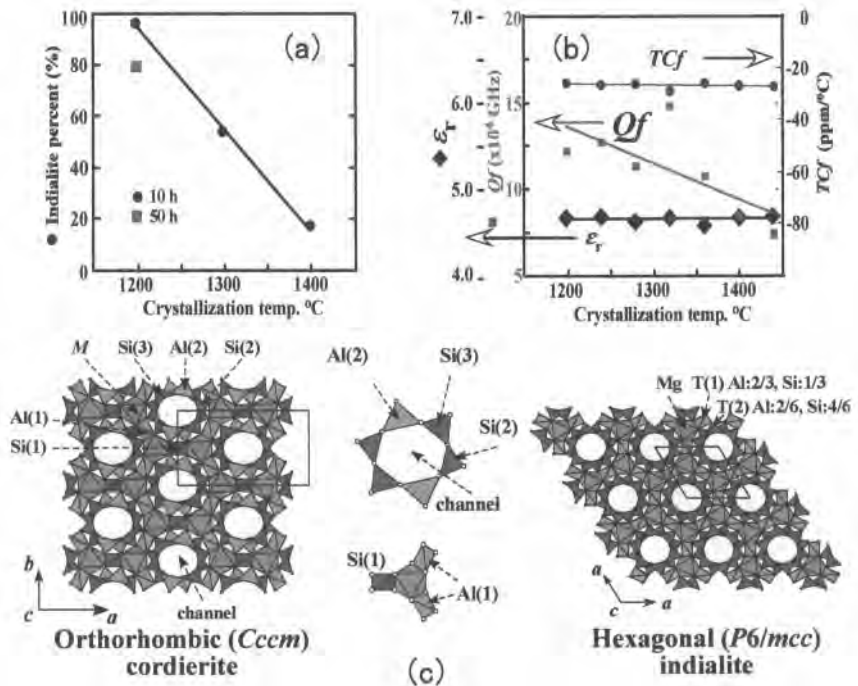


Fig. 4. (a) Indialite percentage sintered for 10 h as a function of temperature, (b) Qf , ϵ_r and TCf of indialite/cordierite glass ceramics sintered at 1200 °C for 10 h as a function of crystallized temperature, (c) polymorph of indialite/cordierite: cordierite is ordered form with orthorhombic, and indialite is disorder form with hexagonal.

(BZN) brings high Q ¹¹⁻¹²⁾. On the other hand, Koga et al.¹³⁻¹⁷⁾ found another factor instead of ordering, and Ohsato et al. presented high symmetry brings high Q instead of ordering on the compounds with order-disorder transition¹⁸⁻²⁰⁾. This conclusion has been derived from following five examples: (1) in the case of BZT, the Q values are not depended on the ordering ratio¹³⁻¹⁵⁾ as shown in Fig. 3(a). (2) In the case of BZN¹⁷⁾ with an clear order-disorder transition at 1350 °C, the low temperature form with high ordering ratio did not show higher Q than the high temperature form with high symmetry as shown in Fig. 3(b). (3) The disordered BZT samples synthesized by spark plasma sintering (SPS) showed the same high Q as ordered ones synthesized by solid state reaction (SSR)¹⁶⁾ as shown in Fig. 3(c). (4) Ni-doped cordierite²¹⁻²³⁾ changing to disordered high temperature form was improved in the Q value. (5) In the case of indialite/cordierite glass ceramics, indialite with high symmetry ($P6/mcc$) shown higher Q than cordierite ($Cccm$) as shown in Fig. 4. Si/AlO₄ tetrahedra of indialite and cordierite are disordered and ordered, respectively²⁴⁾.

The authors resume about the order of crystal structure and microwave dielectric properties. There are two categories of order: one is compositional ordering without order-disorder transition, another is ordering with order-disorder transition. The former case is pseudo-tungstenbronze solid solutions, the latter case is complex perovskite such as BMT, BZT and BZN, as presented above. The ordering of former yield on the same crystal symmetry, but that of the latter on the different symmetry, that is, ordered complex perovskite on the hexagonal and disordered one on the cubic. In the case of complex perovskite with A -site ordering, effect of symmetry might be dominant more than that of ordering.

1-3) High Q by perfect crystal structure

As an example of microwave dielectrics with defect, complex perovskite BZT on the BaO side in the vicinity of BZT as shown in Fig. 5(a) is presented here¹⁵⁻¹⁶⁾. The phase is a single solid solution with disordered structure and has defects in B - and O -sites, as presented by Koga et al. The Qf values of the side become low in order of A, Q, R and S on the \odot line as shown in Fig. 5(b). Kugimiya²⁵⁾ also presented a defect phase in the region $\alpha > 5g/4$ in Ba _{α} Ta _{γ} O _{$\alpha+5g/2$} as shown in Fig. 5(c), the composition denoted by Ba_{1+ α} (Mg_{1/3}Ta_{2/3+ γ} V _{$\alpha-\gamma$})O_{3+ $\alpha+5\gamma/2$} V_{2 $\alpha-5\gamma/2$} has B - and O -site vacancies with holes and electrons.

Though the crystal structural origins of Q factors stated above are intrinsic, extrinsic origin such as impurities, grain growth also degregeate the Q values. The forsterite ceramics are improved from 10,000 to 240,000 GHz by means of using high purity raw materials as shown in Fig. 6(a). The grains of forsterite are very clear and there is no glassy phase among the grains as shown in Fig. 7(c)²⁶⁻²⁹⁾.

Grain growth without rough microstructure sometimes improves Q values as shown in Fig. 7, which shows Qf of Al₂O₃ as a function of grain size. Q of Al₂O₃ was improved from

335,000 to 680,000 GHz by grain growth³⁰⁾. The Qf value of single crystal of Al_2O_3 shows 1,890,000 GHz on // c -axis³¹⁾. So, grain growth might be in the process to single crystal with superior Qf .

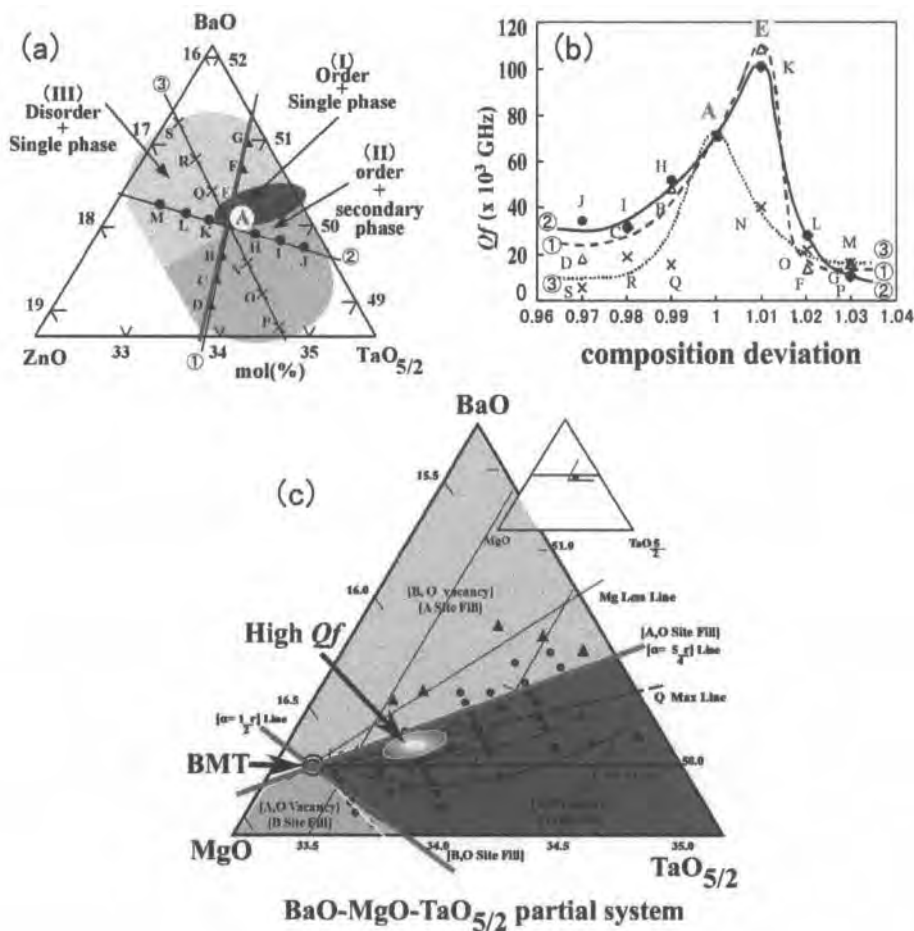


Fig. 5. (a) Partial ternary phase diagram around BZT. Three area are shown as (I) for order/single phase, (II) for order/secondary phase, (III) for disorder/single phase. (b) Qf as a function of composition deviation. On the line ③, disordered single phase region (III) shows low Qf . (c) On the partial ternary system in the vicinity of BMT presented by Kugimiya. Highest Qf value is located near the line of BMT-BaTa_{4/5}O₃.

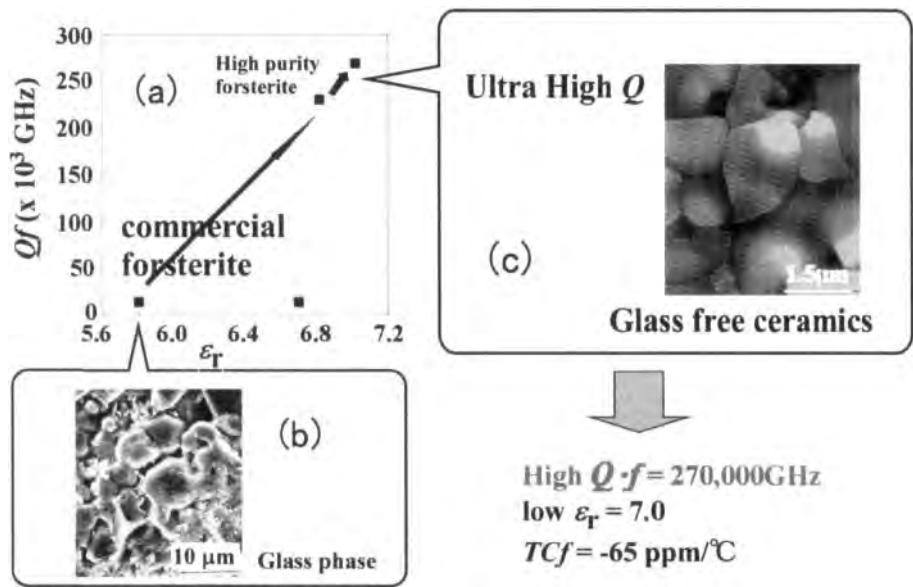


Fig. 6. (a) Qf improved by means of using high purity raw materials. (b) commercial forsterite with glass phase. (c) improved glass free forsterite with high Q .

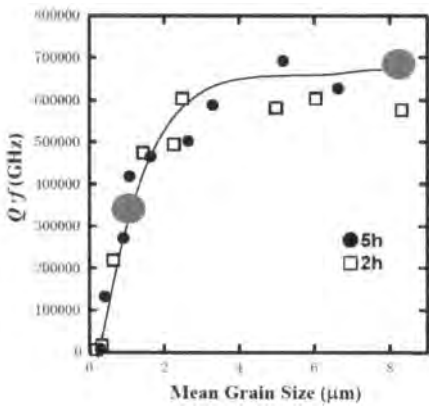


Fig. 7. Qf of Al_2O_3 as a function of mean grain size. Miyauchi improved the Qf of Al_2O_3 from 335,000 to 680,000 GHz.

II) Dielectric constant ϵ_r

Usually, high dielectric constant induces high dielectric loss, because the both properties are proportional to fluctuation of ions. So, Qf values decrease as a function of dielectric constants as shown in Fig. 8(a)³¹⁻³². Pseudo-tungstenbronze solid solutions with high Qf of ca. 10,000 GHz are examples for high dielectric constants of 80-90³⁴. The dielectric constants are almost proportional to the unit cell volumes as shown in Fig. 8(b)³³. This compound has large unit cell volume of over 2,000 Å³ and 1/8 of the unit cell is an asymmetric unit, which space group is *Pnma* (No.62), and multiplicity/Wyckoff letter is $8d^{(1)}$. The large dielectric constants are produced from the 4.5 TiO₆ octahedra in an asymmetric unit as shown in Fig. 8(c).

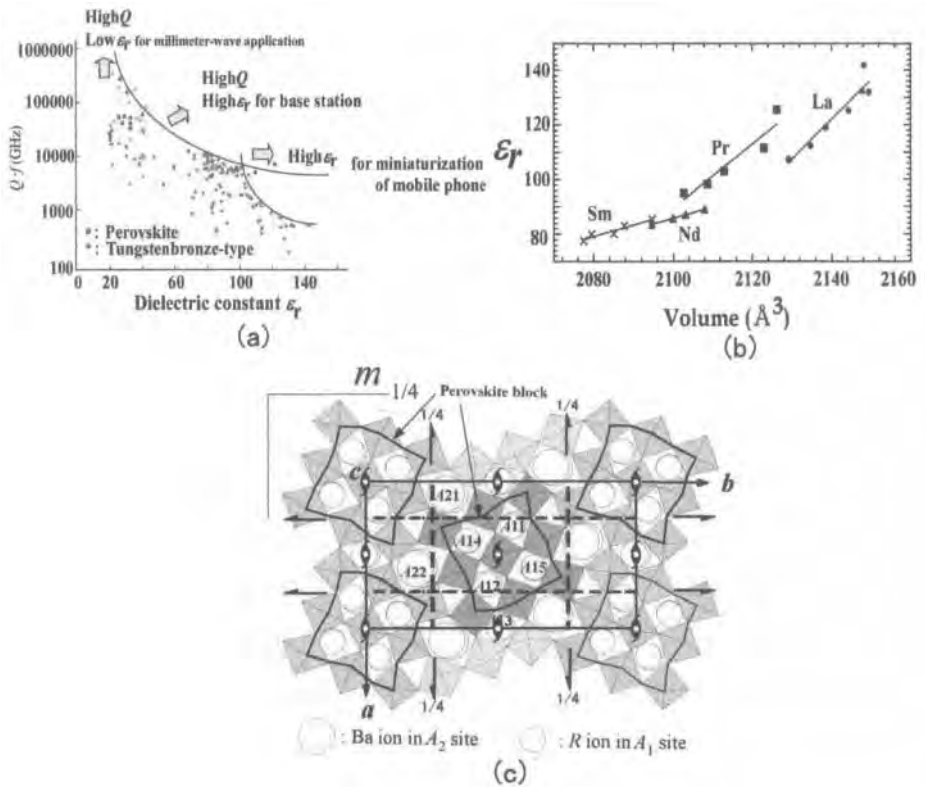


Fig. 8. (a) Qf as a function of dielectric constants, (b) ϵ_r as a function of unit cell volume of the pseudo-tungstenbronze. (c) Large asymmetric unit ($a/2 * b/2 * c/2$) of the pseudo-tungstenbronze.

The TiO_6 octahedra of the pseudo-tungstenbronze structure are tilting forming super structure of two times of c -axis¹⁾. The tilting affects to dielectric constant. Fig. 9 and Table 1 show schematic figures and angles of tilting from c -axis, respectively. The tilting angles are depending on the composition and R cations. As the composition x of 0.5 has much more cations in A_2 sites than $x = 0.7$, the tilting angle decreases and dielectric constant increases. TiO_6 octahedron reached the straight to the c -axes that is the tilting angle of 0° yields large dielectric constants.

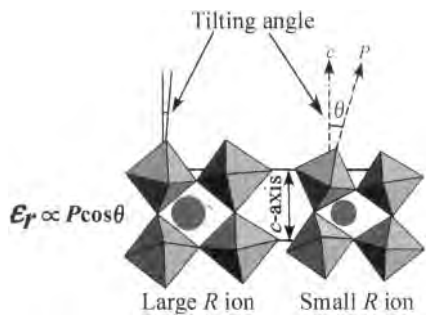


Fig. 9. Corelationship between dielectric constant and tilting angle.

Table 1. Tilting angles θ from c -axis on $x = 0.5$ and 0.7 for $\text{Ba}_{6-3x}\text{R}_{8+2x}\text{Ti}_{18}\text{O}_{54}$.

X	0.5	0.7
Sm	9.990°	10.63°
Nd	9.687°	8.961°

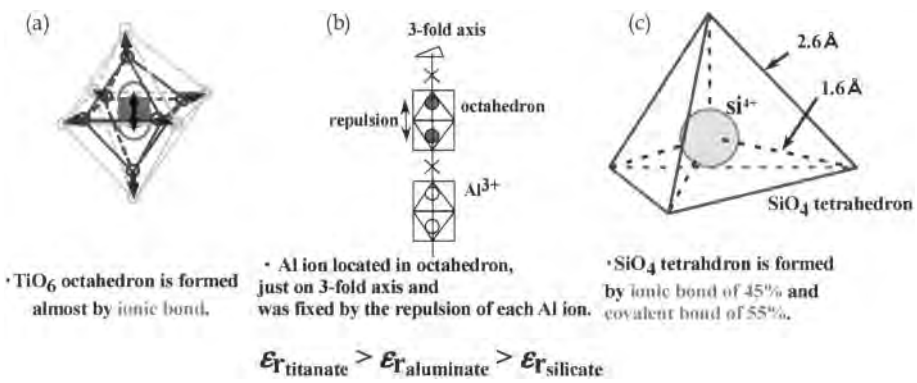


Fig. 10. Dielectric constant due to crystal structure: TiO_6 octahedron (a), octahedra of Al_2O_3 (b), and SiO_4 tetrahedron (c).

Rattling of the cations also affects the dielectric constant. Fig. 10 compared with the effects of rattling on the titanate with TiO_6 octahedron, Al_2O_3 and silicates³⁵. As the titanates have large rattling effect because of large space around Ti cation as shown in Fig. 10(a). As silicates composed by SiO_4 tetrahedron which combined a half by covalency, the cation of Si was bonded hard as shown in Fig. 10(c). So, as the rattling effects are reduced, the dielectric constant is small, which is suitable for millimeter wave dielectrics. The dielectric constant of rutile TiO_2 is about 10, although the structure composed by TiO_6 octahedra. As the octahedra of Al_2O_3 occupied by 2/3 by Ti atoms as shown in Fig. 10(b), Ti ions occupied two adjacent octahedra repulse to the corner of octahedron each other. So, the rutile shows medium dielectric constants.

III) Temperature coefficient of resonant frequency (TCf)

The TCf is defined as following equation:

$$TCf = (f_T - f_{ref}) / f_{ref} (T - T_{ref}) \text{ ppm}/^\circ\text{C} \quad (1)$$

Here, f_T and f_{ref} (GHz) are resonant frequencies on the temperature T , and the reference temperature T_{ref} , respectively. $T_{ref} = -40^\circ\text{C}$ and $T = 85^\circ\text{C}$ on JIS R 1627-1996³⁶. The TCf is very difficult to estimate the value, but it is depending on the crystal structure. Reaney *et al.*³⁷ presented the relationship between temperature coefficient of dielectric constant ($TC\epsilon$) and tolerance factor t , given by eq. (2) on the complex perovskite as shown in Fig. 11³⁸.

$$t = (R_A + R_O) / \sqrt{2} (R_B + R_O) \quad (2)$$

Here, R_A , R_B and R_O are ionic radii of A, B and O ions on ABO_3 perovskite, respectively. There is a relationship between TCf and $TC\epsilon$ given by eq. (3).

$$TCf = -[\alpha + TC\epsilon/2] \quad (3)$$

Here, α is a coefficient of thermal expansion. The t is affected by tilting of octahedron: the octahedron is untilted, in the range of t between 1.055 and 0.985, antiphase tilted in 0.964 and 0.985, and in phase and antiphase tilted in 0.964 and 0.92 as shown in Fig. 11. BMT and $\text{Sr}(\text{Mg}_{1/3}\text{Nb}_{2/3})\text{O}_3$ (SMN) located near zero $TC\epsilon$ at $t = 1.033$ and 0.964, respectively. Solid solutions such as $\text{Ba}_x\text{Sr}_{1-x}(\text{Mg}_{1/3}\text{Ta}_{2/3})\text{O}_3$ (BSMT) also show the same trend. In the case of pseudo-tungstenbronze with tilted octahedra, TCf of Sm-compound is minus as opposite to Nd-, Pr-, and La-compounds with plus TCf . Sm-Nd- and Sm-La-pseudo-tungstenbronze solid solutions yield near zero TCf ³⁹. As seen in these examples, the resonant frequencies are affected by tilting of octahedra. On the other hand, usually, near zero TCf achieved adding different compound with opposite sign of TCf , and the adding ratio is in inverse proportion to the TCf . The TCf should be considered on the resonance which is affected by tilting of octahedron, and volume of additional compound for reducing of TCf .

REFERENCES

- ¹H. Ohsato, "Science of tungstenbronze-type like $\text{Ba}_{6-3x}\text{R}_{8+2x}\text{Ti}_{18}\text{O}_{54}$ ($\text{R}=\text{rare earth}$) microwave dielectric solid solutions", *J. Euro. Ceram. Soc.* **21**, 2703-2711 (2001).
- ²H. Ohsato, Y. Futamata, H. Sakashita, N. Araki, K. Kakimoto and S. Nishigaki, "Configuration and coordination number of cation polyhedra of tungstenbronze-type-like $\text{Ba}_{6-3x}\text{Sm}_{8+2x}\text{Ti}_{18}\text{O}_{54}$ solid solutions", *J. Eur. Ceram. Soc.* **23**, 2529-2533 (2003).
- ³M. T. Sebastian, "Dielectric materials for wireless communication", Elsevier Science Publishers, Amsterdam, 2008.
- ⁴H. Toraya, "Whole-Powder-Pattern Fitting without Reference to a Structural Model: Application to X-ray Powder Diffractometer Data," *J. Appl. Cryst.*, **19**, 440-447 (1986).
- ⁵S. Sasaki, "A Fortran Program for the Least-Squares Refinement of Crystal Structures," *XL Report, ESS*, State Univ. of New York, 1-17 (1982).
- ⁶F. Izumi, and T. Ikeda, "A Rietveld-analysis program RIETAN-98 and its applications to zeolites," *Mater. Sci. Forum*, **321-324**, 198-203, January 2000.
- ⁷B. W. Hakki, and P. D. Coleman, "A Dielectric Resonator Method of Measuring Inductive in the Millimeter Range", *IRE Trans. Microwave Theory & Tech.*, **MTT-8**, 402-410 (1960).
- ⁸Y. Kobayashi, and M. Katoh, "Microwave Measurement of Dielectric Properties of Low-loss Materials by the Dielectric Resonator Method", *IEEE Trans. on MTT-33*, 586-92 (1985).
- ⁹H. Ohsato, M. Imaeda, Y. Takagi, A. Komura and T. Okuda, "Microwave quality factor improved by ordering of Ba and rare-earth on the tungstenbronze-type $\text{Ba}_{6-3x}\text{R}_{8+2x}\text{Ti}_{18}\text{O}_{54}$ ($\text{R}=\text{La}$, Nd and Sm) Solid Solutions", Proceeding of the XIth IEEE International Symposium on Applications of Ferroelectrics, IEEE catalog number 98CH36245, pp509-512 (1998).
- ¹⁰T. Nagatomo, T. Otagiri, "M. Suzuki and H. Ohsato, "Microwave dielectric properties and crystal structure of the tungstenbronze-type like $(\text{Ba}_{1-a}\text{Sr}_a)_6(\text{Nd}_{1-\beta}\text{Y}_\beta)_8\text{Ti}_{18}\text{O}_{54}$ solid solutions", *J. Eur. Ceram. Soc.*, **26**, 1895-1898 (2006.3).
- ¹¹Kawashima, S., Nishida, M., Ueda, I. and Ouchi, H., Dielectric properties at microwave frequencies of the ceramics in $\text{BaOSm}_2\text{O}_3\text{TiO}_2$ system, presented at the 87th Annual Meeting, American Ceramic Society, Cincinnati, OH, May 6, 1985 (Electronics Division Paper No.15-E-85).
- ¹²S. Nomura, K. Toyama and K. Kaneta, *Jpn. J. Appl. Phys.* **21**, 624-626 (1982).
- ¹³E. Koga, H. Moriwake, "Effects of Superlattice Ordering and Ceramic Microstructure on the Microwave Q Factor of Complex Perovskite-Type Oxide $\text{Ba}(\text{Zn}_{1/3}\text{Ta}_{2/3})\text{O}_3$ ", *J. Ceram. Soc. Jpn*, 767-775 (2003) (Japanese).
- ¹⁴E. Koga, H. Moriwake, K. Kakimoto and H. Ohsato, "Influence of Composition Deviation from Stoichiometric $\text{Ba}(\text{Zn}_{1/3}\text{Ta}_{2/3})\text{O}_3$ on Superlattice Ordering and Microwave Quality Factor Q", *J. Ceram. Soc. Jpn.*, 113[2], 172-178 (2005) (Japanese).
- ¹⁵E. Koga, Y. Yamagishi, H. Moriwake, K. Kakimoto and H. Ohsato, "Large Q factor variation within dense, highly ordered $\text{Ba}(\text{Zn}_{1/3}\text{Ta}_{2/3})\text{O}_3$ system ", *J. Euro. Ceram. Soc.*, **26**, 1961-1964 (2006).
- ¹⁶E. Koga, H. Mori, K. Kakimoto and H. Ohsato, "Synthesis of Disordered $\text{Ba}(\text{Zn}_{1/3}\text{Ta}_{2/3})\text{O}_3$ by Spark Plasma Sintering and Its Microwave Q Factor", *Jpn. J. Appl. Phys.*, **45(9B)**, 7484-7488

(2006).

¹⁷E. Koga, Y. Yamagishi, H. Moriwake, K. Kakimoto and H. Ohsato, "Order-disorder transition and its effect on Microwave quality factor Q in $\text{Ba}(\text{Zn}_{1/3}\text{Nb}_{2/3})\text{O}_3$ system", *J. Electroceram*, **17**, 375-379 (2006).

¹⁸H. Ohsato, F. Azough, E. Koga, I. Kagomiya, K. Kakimoto, and R. Freer, "High Symmetry Brings High Q Instead of Ordering in $\text{Ba}(\text{Zn}_{1/3}\text{Nb}_{2/3})\text{O}_3$: A HRTEM Study", *Ceramic Transactions*, Volume 216, 129-136, "Advances in Multifunctional Materials and Systems" Edited by J. Akedo, H. Ohsato, and T. Shimada, Volume Editor: M. Singh, Copyright c 2010 by The American Ceramic Society, Published by John Wiley & Sons, Inc., Hoboken, New Jersey.

¹⁹H. Ohsato, E. Koga, I. Kagomiya and K. Kakimoto, "Origin of High Q for Microwave Complex Perovskite", *Key Eng. Mat.* 421-422 (2010) pp77-80. CSJ Seroes - Publication of the Ceramic Society of Japan - vol.19.

²⁰H. Ohsato, E. Koga, I. Kagomiya and K. Kakimoto, "Phase Relationship and Microwave Dielectric Properties in the Vicinity of $\text{Ba}(\text{Zn}_{1/3}\text{Ta}_{2/3})\text{O}_3$ ", *Ceram. Eng. & Sci. Proc.*, **30** (9), 25-35 (2010).

²¹M. Terada, K. Kawamura, I. Kagomiya, K. Kakimoto and H. Ohsato, "Effect of Ni substitution on the microwave dielectric properties of cordierite", *J. Eur. Ceram. Soc.*, **27**, 3045-3148 (2007.3).

²²H. Ohsato, M. Terada, I. Kagomiya, K. kawamura, K. Kakimoto, and E-S. Kim, "Sintering Conditions of Cordierite for Microwave/Millimeterwave Dielectrics", *IEEE* **55**(5), 1082-1085 (2008).

²³H. Ohsato, I. Kagomiya, M. Terada, and K. Kakimoto, "Origin of improvement of Q based on high symmetry accompanying Si-Al disordering in cordierite millimeter-wave ceramics", *J. Eur. Ceram. Soc.*, **30**, 315-318 (2010).

²⁴H. Ohsato, J.-S. Kim, A.-Y. Kim, C.-I. Cheon, and K.-W. Chae, "Millimeter-Wave Dielectric Properties of Cordierite/Indialite Glass Ceramics", *Jpn. J. Applied Physics*, **50**(9), (2011) 09NF01-1-5.

²⁵K. Kugimiya, Crystallographic study on the Q of $\text{Ba}(\text{Mg}_{1/3}\text{Ta}_{2/3})\text{O}_3$ dielectrics. Abstract for Kansai branch academic meeting held at Senri-Life Science, on the Ceramic Soc Jpn. 2003/9/5; B-20: 20; Abstract for the 10th Meeting of Microwave/Millimeterwave Dielectrics and Related Materials on the Ceram Soc. Jpn. Nagoya Institute of Technology. Japan. 2004/6/21.(Japanese)

²⁶M. Andou, T. Tsunooka, Y. Higashida, H. Sugiura, and H. Ohsato, "Development of high Q forsterite ceramics for high-frequency", applications. MMA2002 Conference, 1-3 September 2002, York, UK.

²⁷T. Tsunooka, M. Andou, Y. Higashida, H. Sugiura and H. Ohsato, "Effects of TiO_2 on sinterability and dielectric properties of high- Q forsterite ceramics", *J. Eur. Ceram. Soc.*, **23**(14), 2573-2578, (2003).

²⁸T. Tsunooka, T. Sugiyama, H. Ohsato, K. Kakimoto, M. Andou, Y. Higashida and H. Sugiura, "Development of Forsterite with High Q and Zero Temperature Coefficient t_f for Millimeterwave Dielectric Ceramics", *Key Engineering Materials*, **269**, 199-202 (2004).

²⁹T. Tsunooka, H. Sugiyama, K. Kakimoto, H. Ohsato, and H. Ogawa, "Zero Temperature Coefficient t_f and Sinterability of Forsterite Ceramics by Rutile Addition", *J. Ceram. Soc. Jpn, Suppl.*, **112**, S1637-S1640 (2004).

³⁰H. Ohsato, T. Tsunooka, Y. Ohishi, Y. Miyauchi, M. Ando and K. Kakimoto, "Millimeter-wave dielectric ceramics with high quality factor and low dielectric constant", *J. Korean Ceram. Soc.*, **40**, 4, 350-353 (2003).

- ³¹J. Krupka, K. Derzakowski, M.E. Tobar, J. Hartnett, and R.G. Geyer, Meas. Sci. Tech. 10(1999).
- ³²H. Ohsato, T. Tsunooka, A. Kan, Y. Ohishi, Y. Miyauchi, Y. Tohdo, T. Okawa K. Kakimoto and H. Ogawa, "Microwave-Millimeterwave Dielectric Materials" *Key Eng. Mat.*, **269**, 195-198 (2004).
- ³³H. Ohsato "Research and Development of Microwave Dielectric Ceramics for Wireless Communications" *J. Ceram. Soc. Jpn.*, **113**[11], 703-711 (2005).
- ³⁴H. Ohsato, T. Ohhashi, H. Kato, S. Nishigaki and T. Okuda, "Microwave Dielectric Properties and Structure of the $\text{Ba}_{6-3x}\text{Sm}_{8+2x}\text{Ti}_{18}\text{O}_{54}$ Solid Solutions", *Jpn. J. Appl. Phys.*, **34**, 187-191 (1995).
- ³⁵H. Ohsato "Microwave Materials with High Q and Low Directric Constant for Wireless Communications", *Mater. Res. Soc. Symp. Proc.* **833**, 55-62 (2005).
- ³⁶JIS R 1627-1996
- ³⁷I. M. Reaney, E. L. Colla and N. Setter, "Dielectric and Structural Characteristics of Ba- and Sr-based Complex Perovskites as a Function of Tolerance Factor", *Jpn. J. Appl. Phys.* **33**, 3984-3990 (1994).
- ³⁸H. D. Megaw, "Crystal structure of double oxides of the perovskite type", *Proc. Phys. Soc.*, **58**, 133 (1946).
- ³⁹H. Ohsato, H. Kato, M. Mizuta, S. Nishigaki and T. Okuda, "Microwave Dielectric Properties of the $\text{Ba}_{6-3x}(\text{Sm}_{1-x}R_y)_{8+2x}\text{Ti}_{18}\text{O}_{54}$ ($R = \text{Nd}$ and La) Solid Solutions with Zero Temperature Coefficient of the Resonant Frequency", *Jpn. J. Appl. Phys.*, **34**, 9B, 5413-5417 (1995).